

Handbook Chemical Engineering Calculations

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In an alternate form, the Redlich-Kwong equation is written as $Z = 1 / (1 - h) - (A/B) [h / (1 + h)]$ where $h = b/V = B P / Z$, $B = b/RT$, $A/B = a / b RT 1.5$, and Z , the compressibility factor, is equal to $P V / RT$. Calculation Procedure 1. Calculate the compressibility factor Z .

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PHYSICAL AND CHEMICAL PROPERTIES 1.5 where T_c , P_c , V_c , and Z_c are critical temperature, critical pressure, critical volume, and critical compressibility factor, respectively. Thus, for p-xylene, $T_c = 412.3[0.567 + 0.106 - (0.106)^2] - 1 = 623.0K(661.8 F)$ (literature value is 616.2 K) $P_c = 106.16(0.34 + 1.378) - 2 = 35.97atm(3644 kPa)$ (literature value is 34.7 atm) V_c

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