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In chemistry, reactivity is a measure of how readily a substance undergoes a chemical reaction. The reaction can involve the substance on its own or with other atoms or compounds, generally accompanied by a release of energy. The most reactive elements and compounds may ignite spontaneously or explosively.

Reactivity Definition in Chemistry - ThoughtCo

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The reactivity series - Reactivity series - GCSE Chemistry ...

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Map: Chemistry & Chemical Reactivity (Kotz et al ...

Atoms are rearranged during chemical reactions, and are not lost or gained. Chemical reactions can be

represented using equations. Catalysts speed up reactions without being used up.

Chemical reactions - Types of reaction - KS3 Chemistry ...

Chemical Reaction: A transformation resulting in a change of composition, constitution and/or configuration of a compound (referred to as the reactant or substrate). Reactant or Substrate: The organic compound undergoing change in a chemical reaction. Other compounds may also be involved, and common reactive partners (reagents) may be identified.

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