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02 - Conversion & Reactor Sizing
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Exam 1 Review Reaction Engineering
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Balance Equations **Reactor Sequence:**
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Octave Levenspiel was a professor of the field Chemical engineering at Oregon State University. In this vast and evergreen field his major interests lied in Chemical Reaction Engineering which is one of the core subjects in Chemical Engineering. He published this book which is considered as a Bible for

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1 Chemical reactions 1.1 Rate of reaction
and dependence on temperature We will
once again look at the formation of

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ammonia (NH_3) from nitrogen and hydrogen (see section Chemical equilibrium of the thermodynamics chapter). This reaction follows the equation: $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$ ($\Delta H^\circ = -92 \text{ kJ mol}^{-1}$, $\Delta S^\circ = 192 \text{ J mol}^{-1} \text{ K}^{-1}$) To find the Gibbs free energy of ...

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By sizing a chemical reactor we mean we're either determining the reactor volume to achieve a given conversion or determine the conversion that can be achieved in a given reactor type and size. Here we will assume that we will be given $-r_A = f(X)$

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and F_{A0} . In chapter 3 we show how to find $-r_A = f(X)$. Given $-r_A$ as a function of conversion, $-r_A = f(X)$, one can size any type of reactor.

Elements of Chemical Reaction Engineering

Octave Levenspiel was a professor of

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chemical engineering at Oregon State University. His principal interest was chemical reaction engineering, and he was the author of a major textbook Chemical Reaction Engineering as well as numerous research publications.

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